T. BALLY and S. MASAMUNE: Cyclobutadiene. *Tetrahedron* **36**(3), 343 (1980).

The second last line on p. 348 should read:

... in the larger (4n)-homologs, [12] and [16]annulene.[40]

Scheme 19 on p. 362 should read:

$$
\begin{aligned}
{}^1\psi_1 &= N(|2\bar{2}| + |3\bar{3}|) && \ldots \quad J_{xx} + K_{xy} \\
{}^1\psi_2 &= N(|2\bar{2}| - |3\bar{3}|) && \ldots \quad J_{xx} - K_{xy} \\
{}^1\psi_3 &= N(|2\bar{3}| + |3\bar{2}|) && \ldots \quad J_{xy} + K_{xy} \\
{}^3\psi_{+1} &= |23| \\
{}^3\psi_0 &= N(|2\bar{3}| - |3\bar{2}|) && \ldots \quad J_{xy} - K_{xy} \\
{}^3\psi_{-1} &= |\bar{2}\bar{3}|
\end{aligned}
$$

The two mathematical expressions on the right side of Scheme 20 should read:

$$
{}^{1,3}\psi_i^0 = f(\phi_{2,3}^0)
$$
$$
{}^{1,3}\psi_i' = f(\phi_{2,3}')
$$

The sentence right after Scheme 20 should read:

Authors who use the first set arrive at an energetic ordering of ${}^3\psi^0 < {}^1\psi_2^0 < {}^1\psi_3^0 < {}^1\psi_1^0$, while those using the primed set find ${}^3\psi' < {}^1\psi_3' < {}^1\psi_2' < {}^1\psi_1'$.

Footnote (m) on p. 362 should read (second sentence):

However, they use $|\bar{2}3|$ instead of $|3\bar{2}|$ which brings about an inversion of signs.

The second last line in the third paragraph on p. 367 should read:

... another reactive hydrocarbon, [3]radialene.[131] It can be used ...

P. CAMPS and C. JAIME: A one step synthesis of 2,4-dialkoxybicyclo[3.2.1]octan-8-ones. Stereochemical assignments using the lanthanide NMR shift reagent, $Eu(FOD)_3$. *Tetrahedron* **36**(3), 393–396 (1980).

The experimental concerning compound **5b** was omitted from p. 396.

(1*RS*,2*RS*,4*RS*,5*SR*)-2,4-*Diethoxybicyclo*[3.2.1]*octan*-8-*one*, **5b**. IR (CCl_4), 1750 cm^{-1}. NMR ($CDCl_3$), δ 3,2–3,8 (comp. abs., 6H), 2,55 (broad, 2H), 1,5–2,3 (broad, 6H), 1,15 and 1,17 (triplets, J = 6.5 Hz, 6H). MS 212 (M^+, 3), 166 (M^+—EtOH, 39), 129 (EtO—$\overset{+}{C}$H—CH=CH—OEt, 100), 101 (EtO—$\overset{+}{C}$H—CH=CH—OH, 42), 85 (EtO—$\overset{+}{C}$H—CH=CH_2, 39). (Found: C, 67,90; H, 9.66. $C_{12}H_{20}O_3$ requires: C, 67,89; H, 9,50%).